

SIMPLE PREPARATION OF A USEFUL C/D-RING FRAGMENT
FOR THE CONSTRUCTION OF 11-KETO STEROIDS

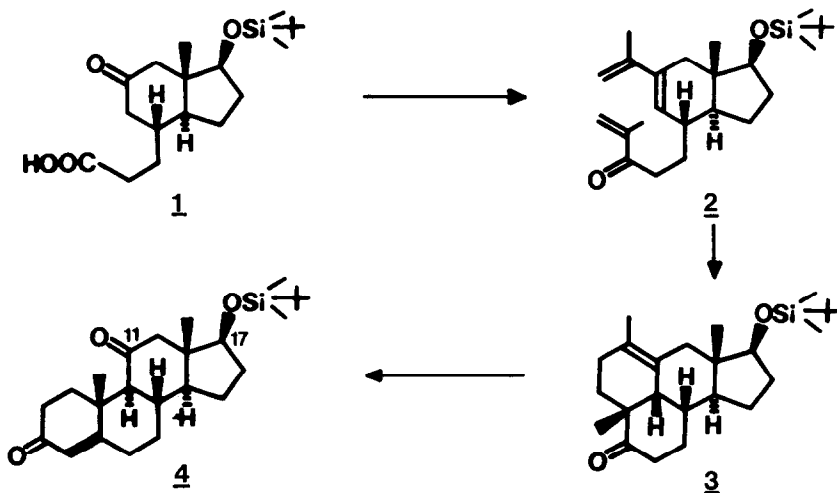
Gilbert Stork*, Glenn Clark and Thomas Weller¹⁾

Department of Chemistry, Columbia University

New York, New York 10027

Summary: Starting with the readily available enol-lactone 5 the indanone-propionic acid 1 is prepared in few steps with 33% overall yield.

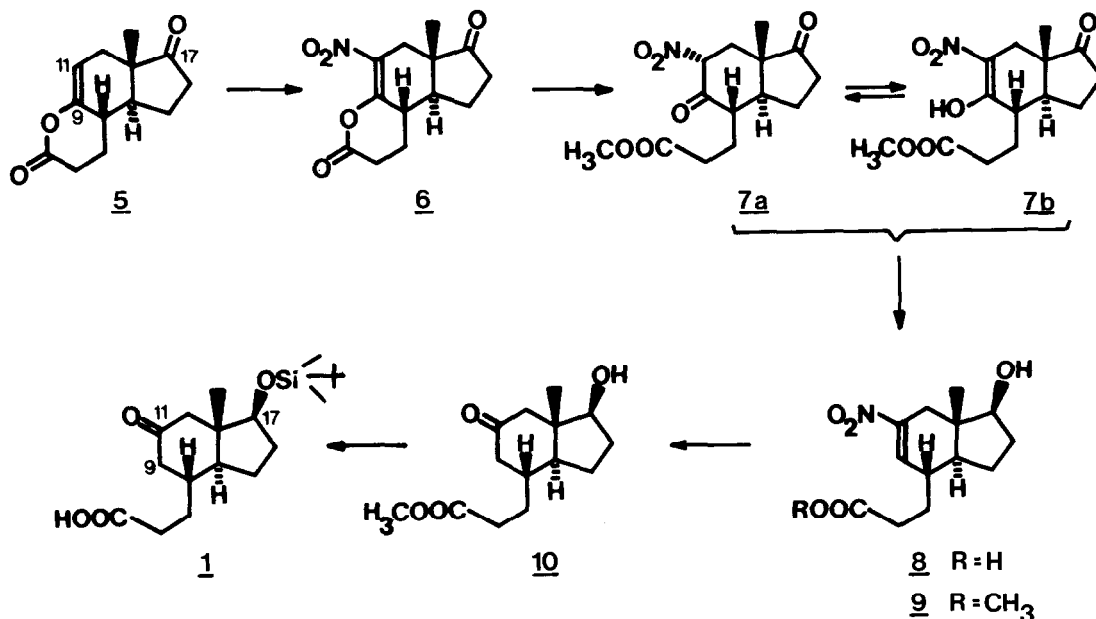
We have recently shown that 11-keto steroids (e.g. 4) can be constructed by an intramolecular Diels-Alder reaction 2 \longrightarrow 3, starting with the C/D-ring fragment 1²⁾.



The ongoing interest in the field of 11-keto steroid synthesis³⁾⁴⁾⁵⁾ prompts us to disclose our approach to the preparation of the keto acid 1.

Our starting material, the enol-lactone 5, is readily available in optically active form by microbial degradation of cheap β -sitosterol⁶⁾⁷⁾.

The problem was thus restricted to finding a concise route for the C(9)-C(11)⁸⁾ carbonyl transposition ($\underline{5} \rightarrow \underline{1}$)⁹⁾. After some experimentation we found that nitration¹⁰⁾¹²⁾ of $\underline{5}$ with concentrated nitric acid in acetic anhydride led to the crystalline nitro enol-lactone $\underline{6}$ in 92% yield. (For the reaction conditions and some properties of the compounds $\underline{1}$ and $\underline{6-10}$ see the accompanying table).



The direct reduction of $\underline{6}$ to the nitro-olefin $\underline{8}$ turned out to be a rather difficult task¹¹⁾, which could not be achieved in a single step. Therefore the enol-lactone $\underline{6}$ was converted to the nitro-ester $\underline{7}$ (91% yield) by simple solvolysis in methanol. Inspection of the ¹H-NMR-spectrum indicated a 1:1-mixture of the keto-form $\underline{7a}$ and the enol-form $\underline{7b}$. The crude mixture $\underline{7a/7b}$ was reduced with sodium borohydride in the presence of $\text{CeCl}_3 \cdot 7\text{H}_2\text{O}$ ¹³⁾ to give the desired 17 β -hydroxy-nitro-olefin $\underline{9}$ (60% yield after flash chromatography¹⁶⁾). Upon treatment of $\underline{9}$ with an aqueous solution of titanium trichloride, buffered by addition of ammonium acetate¹⁴⁾ the 17 β -hydroxy-keto-ester $\underline{10}$ was obtained in 77% yield. Finally, protection of the free hydroxyl group of $\underline{10}$ by formation of the t-butyl-dimethyl-silyl-ether¹⁵⁾ followed by saponification of the methyl ester produced the desired indanone propionic acid $\underline{1}$ in 79% yield.

<p><u>Table.</u> Reaction conditions for the preparation of the compounds <u>1</u> and <u>6-10</u> together with some characteristic spectroscopic data. The $^1\text{H-NMR}$-spectra were run at 270 MHz. (δ-values in ppm with SiMe_4 as internal standard)</p>	
<u>6</u>	from <u>5</u> , 1.1 mol-equiv. 70% HNO_3 , acetic anhydride, -20°C to room temperature overnight, followed by removal of solvents at $70^\circ\text{C}/1$ Torr; IR (CHCl_3): 1780, 1740, 1520, 1330 cm^{-1} ; $^1\text{H-NMR}$ (CDCl_3): 1.10 (s, 3H).
<u>7a/7b</u>	from <u>6</u> , CH_3OH , catalytic amount of p-toluenesulfonic acid, room temperature overnight; IR (CHCl_3): 1730, 1560 cm^{-1} ; $^1\text{H-NMR}$ (CDCl_3): 1.00 and 1.30 (2 s, together 3H), 3.75 (s, 3H), 5.45 (dxd, J=5 and 12 Hz, 0.5H), 14.40 (s, 0.5H); MS: 298 ($\text{M}^{\ddagger}+1$).
<u>9</u>	from <u>7a/7b</u> , 2 mol-equiv. NaBH_4 , 3 mol-equiv. $\text{CeCl}_3 \cdot 7\text{H}_2\text{O}$, EtOH, 2 h at room temperature; IR (CHCl_3): 3640, 1735, 1520, 1340 cm^{-1} ; $^1\text{H-NMR}$ (CDCl_3): 0.79 (s, 3H), 3.69 (s, 3H), 3.89 (t, J=7 Hz, 1H), 7.19 (br.s, 1H).
<u>10</u>	from <u>9</u> , 6 mol-equiv. TiCl_3 in $\text{H}_2\text{O}/\text{CH}_3\text{OH}$ 1:1 in the presence of NH_4OAc ; IR (CHCl_3): 3640, 1730, 1710 cm^{-1} ; $^1\text{H-NMR}$ (CDCl_3): 0.72 (s, 3H), 3.67 (s, 3H), 3.88 (t, J=8.5 Hz, 1H).
<u>1</u>	from <u>10</u> , $\text{Me}_2\text{t-BuSiOSO}_2\text{CF}_3$, 2,6-lutidine, CH_2Cl_2 , -70°C followed by hydrolysis with KOH, CH_3OH , room temperature; IR (CHCl_3): 3300-2400 (br), 1710 cm^{-1} ; $^1\text{H-NMR}$ (CDCl_3): 0.02 (s, 6H), 0.70 (s, 3H), 0.87 (s, 9H), 3.80 (t, J=8 Hz, 1H); m.p. $127-129^\circ\text{C}$.

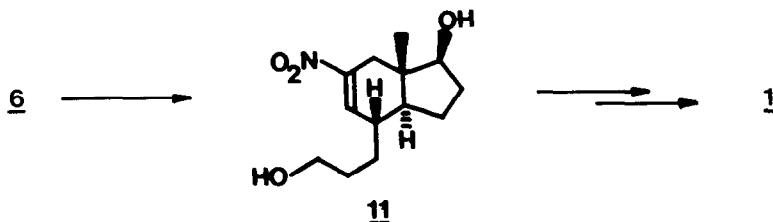
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References and Notes

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- 11) Treatment of 6 with a large excess of sodium borohydride led to the formation of the diol 11¹² (84% yield), which could also be con-



verted to 1 by a different route, although much less efficiently than described above.

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